

## Combined Gas Law Questions

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Note: atmospheric pressure (atm) is  $1 \times 10^5 \text{ Pa}$

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

1. If I initially have a gas at pressure of 12atm, a volume of 23 litres and a temperature of 200K, and then I raise the pressure to 14atm and increase the temperature to 300K, what is the new volume of the gas?
2. A gas takes up a volume of 17 litres, has a pressure of 2.3atm, and a temperature of 299K. If I raise the temperature to 350K and lower the pressure to 1.5atm, what is the new volume of the gas?
3. A gas that has a volume of 28 litres, a temperature of  $45^\circ\text{C}$ , and an unknown pressure has its volume increased to 34 litres and its temperature decreased to  $35^\circ\text{C}$ . If I measure the pressure after the change to be 2.0atm, what was the original pressure of the gas?
4. A gas has a temperature of  $14^\circ\text{C}$ , and a volume of 4.5 litres. If the temperature is raised to  $29^\circ\text{C}$  and the pressure does not change, what is the new volume of the gas?
5. If I have 17 litres of gas at a temperature of  $67^\circ\text{C}$  and a pressure of 88.89atm, what will be the pressure of the gas if I raise the temperature to  $94^\circ\text{C}$  and decrease the volume to 12 litres?
6. I have an unknown volume of gas at a pressure of 0.5atm and a temperature of 325K. If I raise the pressure to 1.2 atm, decrease the temperature to 320K, and measure the final volume to be 48 litres, what was the initial volume of the gas?
7. If I have 21 litres of gas held at a pressure of 78 atm and a temperature of 900K, what will be the volume of the gas if I decrease the pressure to 45 atm and decrease the temperature to 750K?
8. If I have 2.9L of gas at a pressure of 5atm and a temperature of  $50^\circ\text{C}$ , what will be the temperature of the gas if I decrease the volume of the gas to 2.4L and decrease the pressure to 3atm?
9. I have an unknown volume of gas held at a temperature of 115K in a container with a pressure of 60atm. If by increasing the temperature to 225K and decreasing the pressure to 30atm causes the volume of the gas to be 29 litres, how many litres of gas did I start with?

# Gas Law Essay Questions

**Jaramillo Cayenne Shepard**

## Gas Law Essay Questions:

The Ultimate Bar Exam Playbook Jaramillo Cayenne Shepard, 2024-12-30 The Bar Exam Blueprint is your ultimate guide to mastering the bar exam in just 30 days Designed for law students graduates and repeat takers this action oriented book combines proven strategies expert insights and practical tools to help you conquer the MBE essays and performance tests with confidence Inside you ll find Step by step daily plans to guide your study schedule and keep you on track Simplified subject outlines for heavily tested areas like Contracts Torts and Evidence Effective essay writing techniques using the IRAC method to ensure clarity and precision Performance test strategies to help you manage time organise tasks and excel under pressure Practice questions and sample answers to refine your understanding and application of the law Stress management tools and tips to stay calm focused and prepared on exam day This comprehensive book is packed with memory aids templates and checklists to streamline your preparation and maximise efficiency It s not just a study guide it s a complete system designed to help you succeed no matter where you are in your bar exam journey With The Bar Exam Blueprint you ll gain the confidence skills and strategies needed to pass the bar and achieve your dream of becoming a licensed attorney Start your 30 day mastery plan today and take the first step toward bar exam success

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