

Isotope	Synthetic Notation	Number of Protons	Number of Electrons	Number of Neutrons
11. Hydrogen-2		1	1	1
12.	$^{14}_{12}\text{C}$			
13.		8		18
14. Copper-63				34
15.	$^{238}_{92}\text{U}$			

Section Early Theories of Matter

In your textbook, read about the philosophers John Dalton, and defining the atom. For each statement below, write *true* or *false*.

1. Ancient philosophers regularly performed controlled experiments.
2. Philosophers formulated explanations about the nature of matter based on their own experiences.
3. Both Democritus and Dalton suggested that matter is made up of atoms.
4. Dalton's atomic theory stated that atoms separate, combine, or rearrange in chemical reactions.
5. Dalton's atomic theory stated that matter is mostly empty space.
6. Dalton was correct in thinking that atoms could not be divided into smaller particles.
7. Dalton's atomic theory stated that atoms of different elements combine in simple whole-number ratios to form compounds.
8. Dalton thought that all atoms of a specific element have the same mass.
9. Democritus proposed that atoms are held together by chemical bonds, but no one believed him.
10. Dalton's atomic theory was based on careful measurements and extensive research.
11. There are no instruments powerful enough to magnify atoms so that they can be seen.
12. The smallest particle of an element that retains the properties of that element is called an atom.

Section How Atoms Differ

In your textbook, read about atomic number.

For each statement below, write *true* or *false*.

1. The number of neutrons in an atom is referred to as its atomic number.
2. The periodic table is arranged by increasing atomic number.
3. Atomic numbers are equal to the number of electrons in an atom.
4. The number of protons in an atom identifies it as an atom of a particular element.
5. Most atoms have either a positive or a negative charge.

Answer the following questions.

6. Lead has an atomic number of 82. How many protons and electrons does lead have?
7. Oxygen has 8 electrons. How many protons does oxygen have?
8. Zinc has 30 protons. What is its atomic number?
9. Astatine has 85 protons. What is its atomic number?

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