

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a heterogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution and an unsaturated solution.

Define the term solvent and provide an example.

What is solubility, and how is it measured?

Describe the difference between concentrated and dilute.

What are the properties of acids?

What are the properties of bases?

What is the pH scale?

What is the difference between strong acids and weak acids?

What is the difference between strong bases and weak bases?

What is the difference between a strong acid and a weak acid?

What are the properties of bases and how are they measured?

Explain the significance of the pH scale.

How does the concentration of an acid affect its pH?

What are the properties of acids during neutralization?

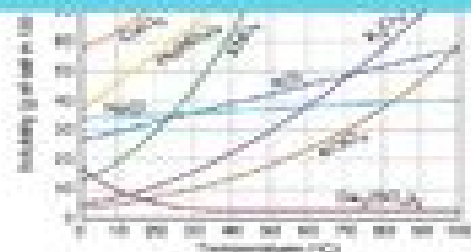
What are the properties of strong acids and weak acids?

What is the difference between strong acids and weak acids?

What is the difference between strong acids and weak acids?

What is the difference between strong acids and weak acids?

What is the difference between strong acids and weak acids?



20. Describe how you would use the graph to determine the amount of Na_2CO_3 that can be dissolved in 100 grams of water at 60°C .

22. Which salt has the highest solubility at 80°C ?



ANSWER KEY INCLUDED

Glencoe Acids Bases Study Guide Answer Key

John S. Phillips, Cheryl Wistrom



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